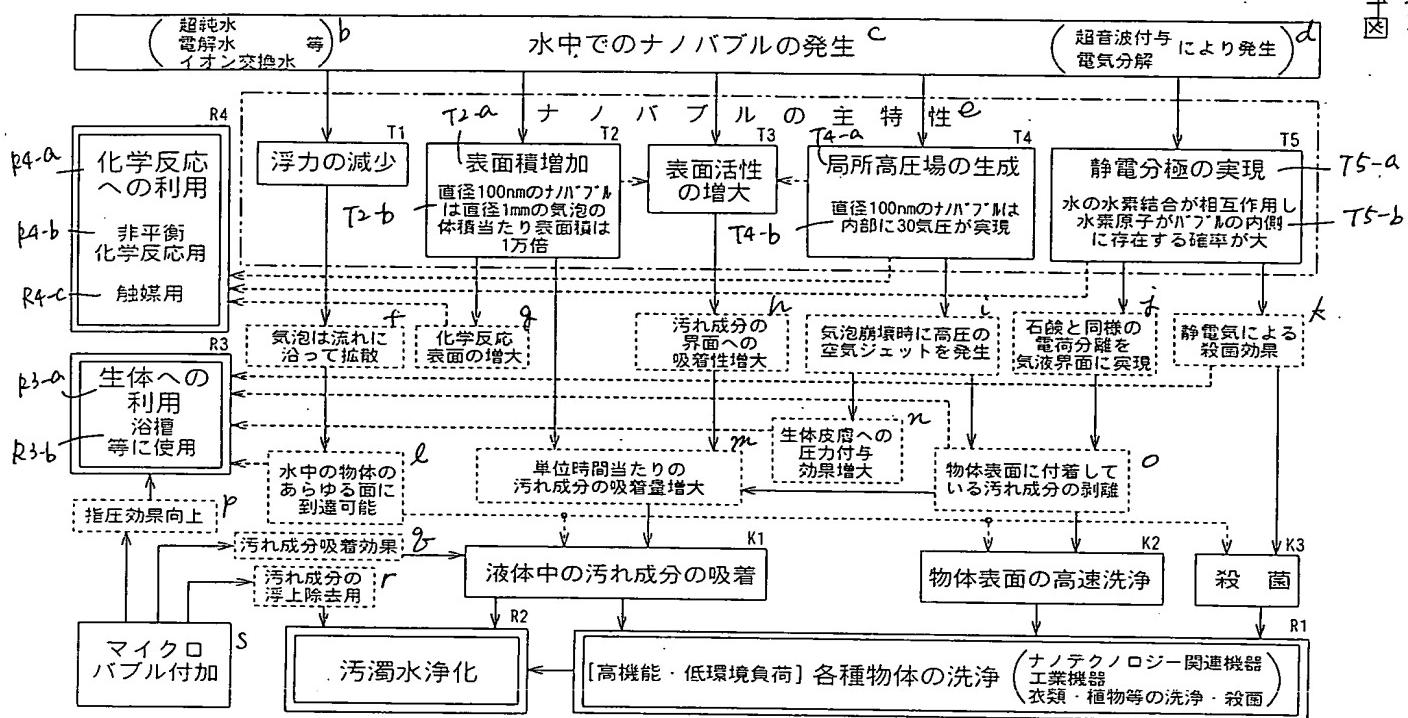


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第十図

ナノバブル利用技術



- a. NANOBUZZLE UTILIZATION TECHNOLOGY
- b. (ULTRA-PURE WATER, ELECTROLYZED WATER, ION-EXCHANGED WATER, AND SO FORTH)
- c. GENERATION OF NANOBUBBLES IN WATER (GENERATION BY APPLICATION OF ULTRASONIC WAVE AND ELECTROLYSIS)
- d. MAIN FEATURES OF NANOBUBBLE
- e. USE FOR CHEMICAL REACTION
- f. APPLICATION TO NONEQUILIBRIUM CHEMICAL REACTION
- g. FOR CATALYST
- h. DECREASE OF BUOYANCY FORCE
- i. INCREASE OF SURFACE AREA
- j. SURFACE AREA OF NANOBUBBLE THAT OF AIR BUBBLE OF 1 mm DIAMETER IS 10,000 TIMES
- k. INCREASE OF SURFACE ACTIVITY
- l. GENERATION OF LOCAL HIGH-PRESSURE FIELD
- m. PRESSURE IN NANOBUBBLE OF 100 nm DIAMETER IS 30 ATMS
- n. ESTABLISHMENT OF ELECTROSTATIC POLARIZATION
- o. PROBABILITY THAT HYDROGEN ATOMS ARE PRESENT INSIDE BUBBLE IS HIGH BECAUSE HYDROGEN BONDS OF WATER INTERACT WITH ONE ANOTHER
- p. USE FOR ORGANISM
- q. USE FOR BATHTUB AND SO FORTH
- r. BUBBLES DIFFUSE ALONG STREAM
- s. INCREASE OF CHEMICAL REACTION SURFACE
- t. INCREASE OF ADSORPTIBILITY OF FOUL COMPONENTS ON INTERFACE
- u. HIGH-PRESSURE AIR JET IS FORMED WHEN BUBBLE COLLAPSES
- v. CHARGE SEPARATION IS REALIZED AT AIR-LIQUID INTERFACE LIKE THAT BY SOAP
- w. STERILIZATION EFFECT BY STATIC ELECTRICITY CAN REACH ANY SURFACE OF OBJECT IN WATER
- x. INCREASE OF AMOUNT OF FOUL COMPONENTS ADSORBED PER UNIT TIME
- y. ENHANCEMENT OF EFFECT OF EXERTING PRESSURE ON ORGANISM SKIN
- z. SEPARATION OF FOUL COMPONENTS ADHERING TO OBJECT SURFACE
- aa. ENHANCEMENT OF ACUPRESSURE EFFECT
- bb. FOUL COMPONENT ADSORPTION EFFECT
- cc. FOR FLOTATION REMOVAL OF FOUL COMPONENTS
- dd. ADSORPTION OF FOUL COMPONENTS IN LIQUID
- ee. HIGH-SPEED CLEANING OF OBJECT SURFACE
- ff. STERILIZATION
- gg. IMPARTING OF MICROBUBBLES
- hh. PURIFYING OF POLLUTED WATER
- ii. VARIOUS OBJECTS (CLEANING AND STERILIZATION OF NANOTECHNOLOGY-ASSOCIATED EQUIPMENT, INDUSTRIAL EQUIPMENT, CLOTHES, PLANTS, AND SO FORTH)

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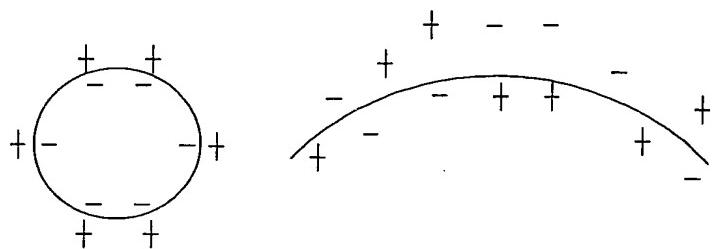
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Fig. 2
第2図

ELECTROLYTIC SEPARATION PHENOMENON
SIMILAR TO SOAP ON NANOBUBBLE SURFACE

ナノバブルの表面における
石鹼類似の電解分離現象



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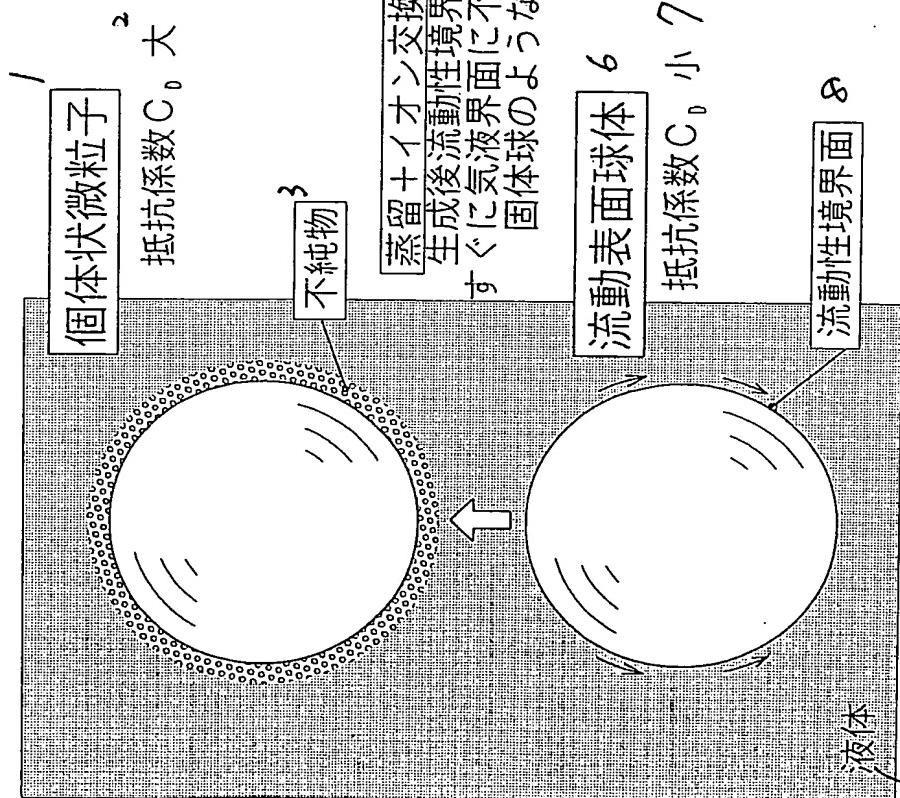
FIG. 3

図 3

5

蒸留 + イオン交換水の水質
比抵抗

10 MΩ・cm 程度
微粒子数 (粒径 $> \phi 0.5 \mu\text{m}$)
10000 個/m³
TOC (全有機炭素量)
1 ppm 程度



- SPECIFIC RESISTANCE: ABOUT 10 MΩ•cm
FINE PARTICLES NUMBER (GRAIN DIAMETER $> \phi 0.5 \mu\text{m}$): 10,000/m³
TOC (TOTAL ORGANIC CARBON): ABOUT 1 PPM
6. LIQUIDITY BOUNDARY SURFACE SPHERE
7. COEFFICIENT OF RESISTANCE C_D : SMALL
8. LIQUIDITY BOUNDARY SURFACE
9. LIQUID

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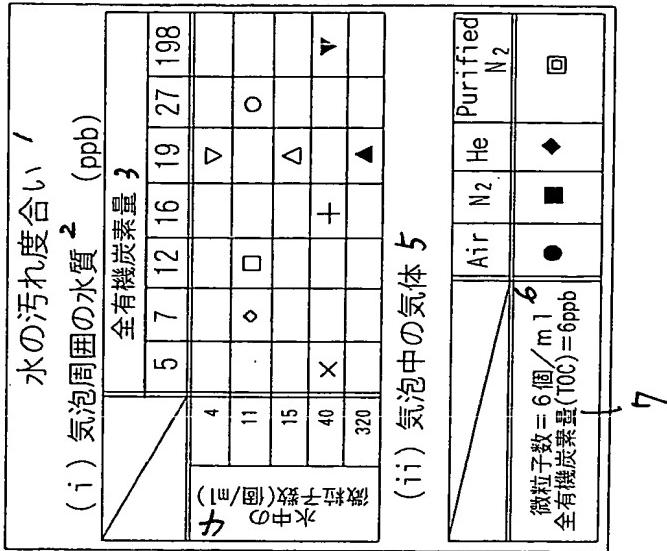
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F/G.4

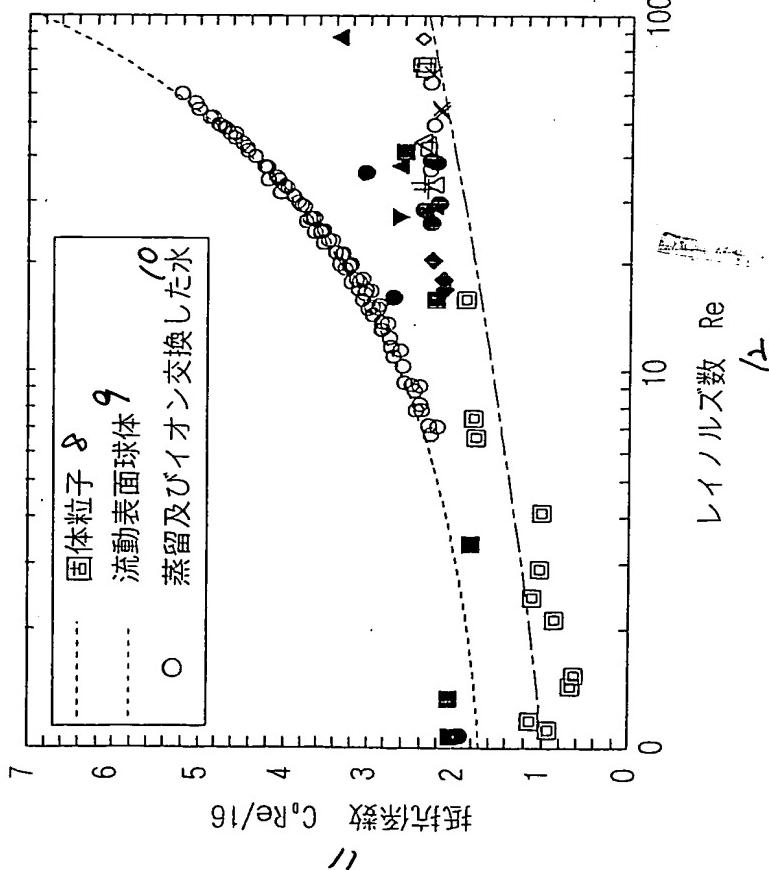
第4回

1. DEGREE OF POLLUTION OF WATER
2. WATER QUALITY OF PERIPHERY OF BUBBLE
3. TOTAL ORGANIC CARBON
4. NUMBER OF FINE PARTICLES (NUMBER/ML) WITHIN WATER
5. GAS WITHIN BUBBLE
6. NUMBER OF FINE PARTICLES = 6/ml
7. TOTAL ORGANIC CARBON
8. SOLID PARTICLE
9. LIQUIDITY BOUNDARY SURFACE SPHERE
10. WATER MIXED WITH PURE WATER AND ION EXCHANGED WATER
11. RESISTANCE COEFFICIENTS
12. REYNOLDS NUMBERS

(a)

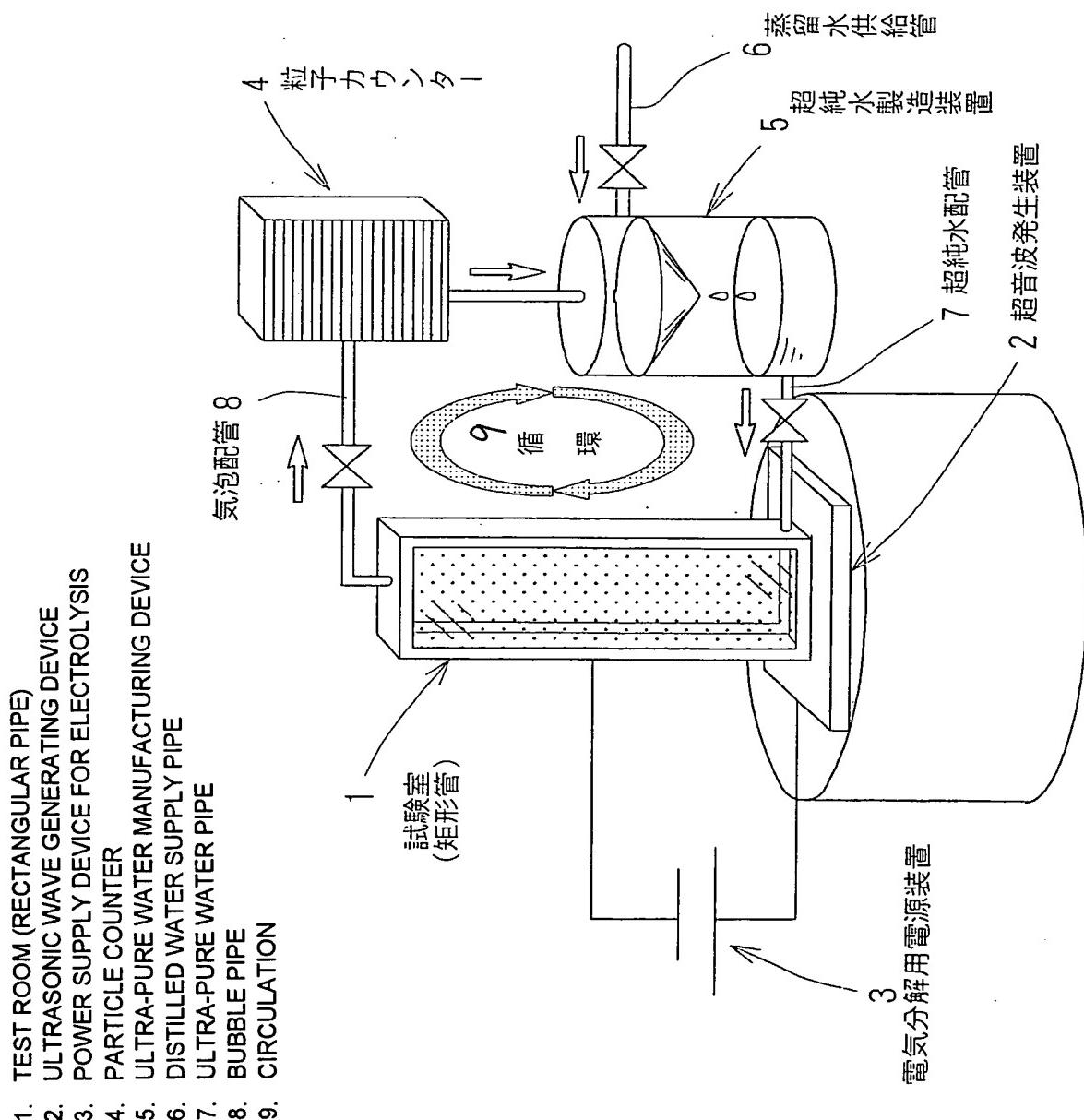


(b)



F(6.5)

第5図



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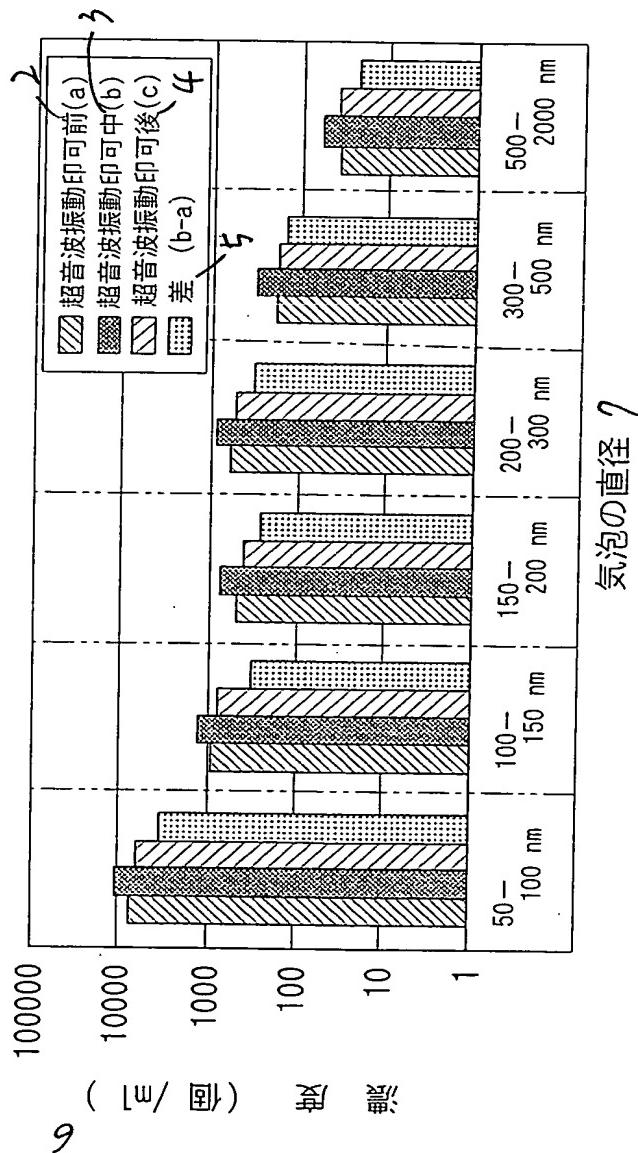
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第6回

超音波振動印可前と印可中の気泡の濃度差 ($\gamma = 2.0$)

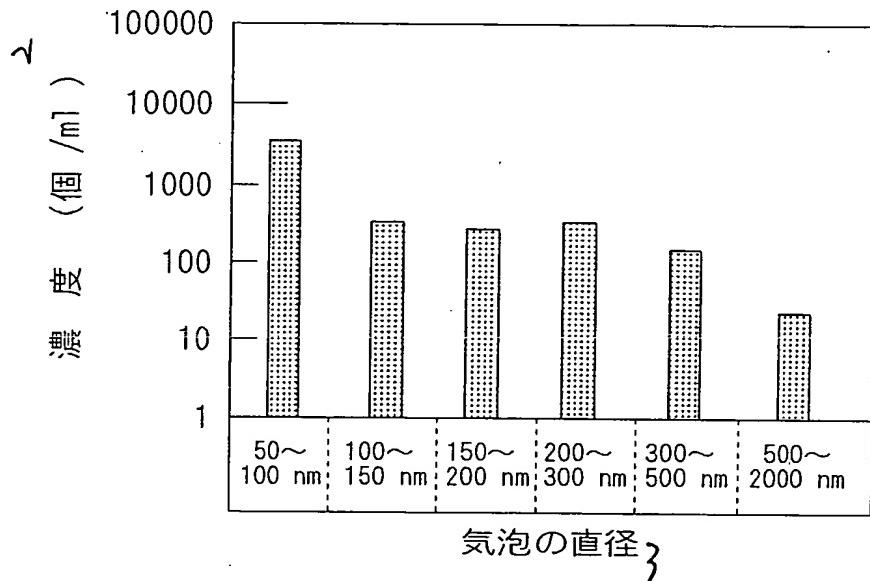


1. DIFFERENCE OF DENSITIES OF BUBBLES BETWEEN BEFORE APPLICATION OF ULTRASONIC WAVE VIBRATION AND DURING APPLICATION THEREOF
2. BEFORE APPLICATION OF ULTRASONIC WAVE VIBRATION
3. DURING APPLICATION OF ULTRASONIC WAVE VIBRATION
4. AFTER APPLICATION OF ULTRASONIC WAVE VIBRATION
5. DIFFERENCE (B-A)
6. DENSITY (NUMBER/ml)
7. DIAMETER OF BUBBLE

FIG.7

第7図

超音波印可中のナノバブルの発生濃度



1. DENSITY OF NANOBUBBLES GENERATED DURING APPLICATION OF ULTRASONIC WAVE
2. DENSITY (NUMBER/ml)
3. DIAMETER OF BUBBLE

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